



## Bridging the Gap Between Lab Scale and Full Scale Catalysis Experimentation

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### Introduction

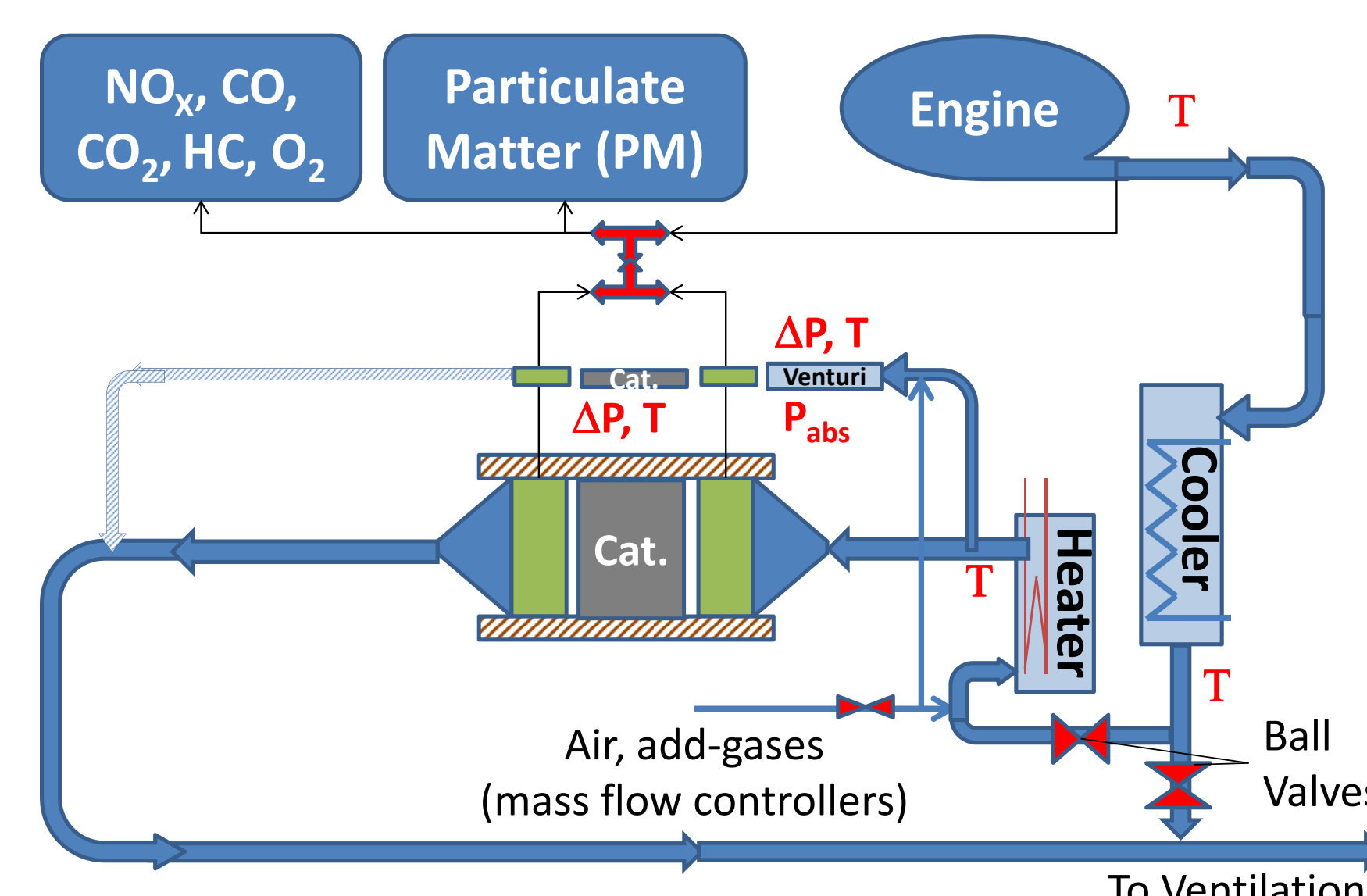
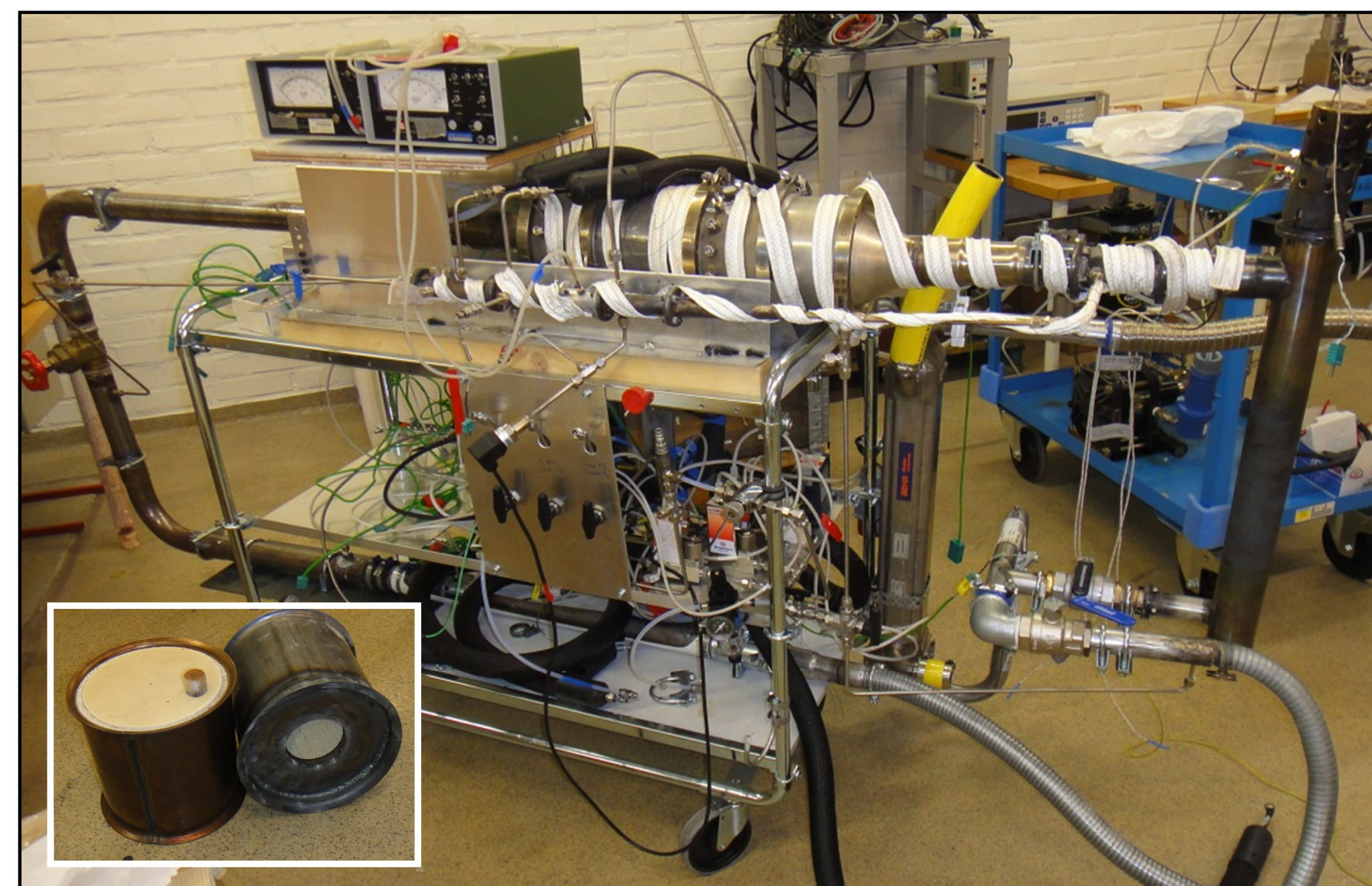
The knowledge transfer between lab scale and full scale is becoming increasingly important.

Lab scale understanding of complex catalytic phenomena needs to be transferred to full scale with high precision.

Real world issues need to be revisited in more fundamental studies

### Objective

To develop a reactor system (EATS, Emission Aftertreatment System) that creates a bridge between lab scale and full scale research experimentation.



### EATS experimental features

#### Highly flexible

- Different engine type, size
- Two "legs" (0.2-20 l/min or 10-1000 l/min)
- Different catalyst sizes

#### Independent residence time

- Set by flow by-pass (ball valves)
- Measured by pressure drop or venturimeter

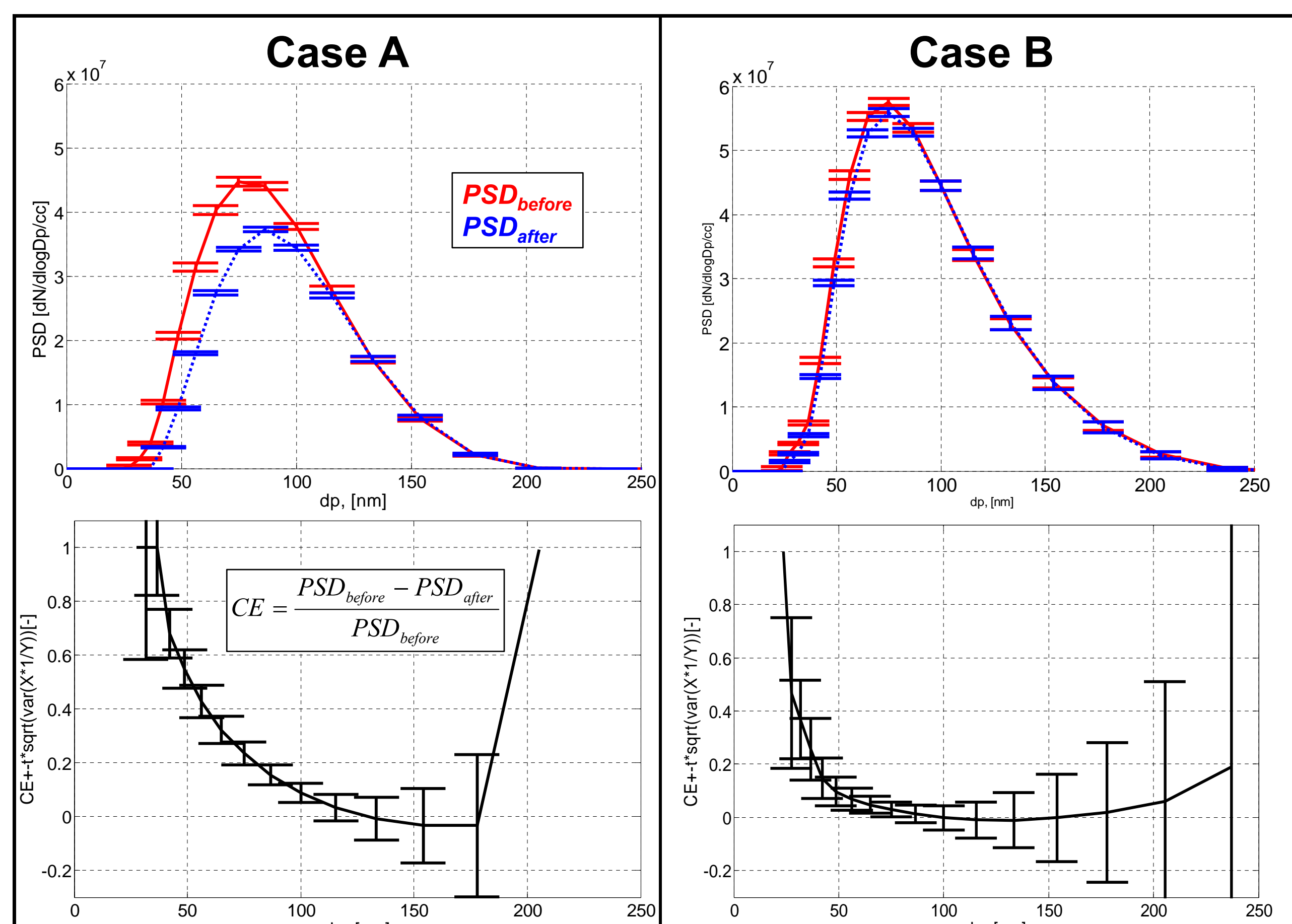
#### Independent concentrations

- Added air or synthetic gases
- Mass flow controllers (Labview)

#### Independent temperatures

- Cooler & heater
- "Adiabatic" heating tape

### Example 1: Particulate Matter (PM) Capture



Engine: 5 cyl (2.4 dm <sup>3</sup> ), Low sulfur diesel fuel (MK1), 1200 rpm, 30 Nm (BMEP=1.57 bar)					
Substrate: Bare ceramic monolith (5.66" x 6", 400 cpsi, 2.5 dm <sup>3</sup> )					
PM instrument: DMS 500, 1dil=1:1, 2dil=100:1, T <sub>sample line</sub> =75 °C, P <sub>instr.</sub> =0.25 bar					
PM capture conditions	T <sub>cat</sub> [°C]	T <sub>min</sub> /T <sub>max</sub> [°C]	Flow [Nl/min]	Re <sub>channel</sub> [-]	N (data points before/after)
Case A	222	194/258	66	2.7	185/197
Case B	157	154/159	210	10.8	62/110

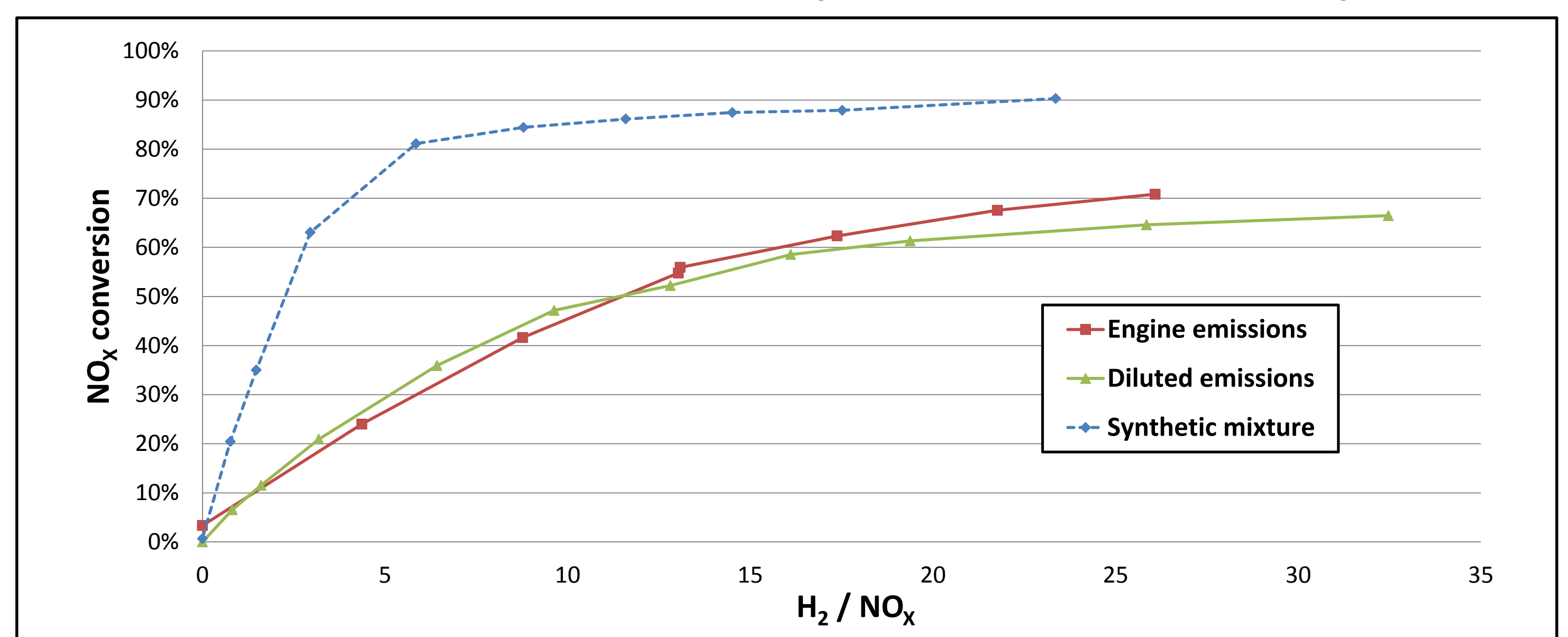
### Results PM Capture

- Low flow required to get significant capture
- Capture efficiency (CE) depend on reactor conditions (T,Q)
- Uncertainty for CE depends on signal level as well as secondary effects (PM, flow properties, sampling procedures)

### Significance

- Important to measure accurately and representatively
- Secondary effects apparent when comparing with simulations (on-going work)

### Example 2: H<sub>2</sub>-assisted NH<sub>3</sub>-SCR over Ag-Al<sub>2</sub>O<sub>3</sub>



Engine: 1 cyl (0.48dm <sup>3</sup> ), Diesel fuel (MK1), 1200 rpm, 14.7 Nm (BMEP=3.85 bar)							
Catalyst: 2x2 cm (6.28 cm <sup>3</sup> ), Ag/Al <sub>2</sub> O <sub>3</sub> : 2 %w/w (freeze-dried Sol-Gel preparation)							
Add-gases: NH <sub>3</sub> (4 % in N <sub>2</sub> ), H <sub>2</sub> (40 % in He), compressed dry air for diluted emissions and NO (900 ppm in N <sub>2</sub> ) for synthetic mixture							
Inlet conditions	T <sub>cat</sub> [°C]	Flow [Nl/min]	NO <sub>x</sub> [ppm]	NH <sub>3</sub> [ppm]	H <sub>2</sub> O [%]	CO [ppm]	CO <sub>2</sub> [%]
Engine emissions	255	1.6	548	451	3.5	150	4.2
Diluted emissions	256	1.6	376	264	3.0	102	2.8
Synthetic mixture	252	0.6	293	234	0	0	0

### Results Ag-Al<sub>2</sub>O<sub>3</sub> SCR

- Same H<sub>2</sub> effect upon dilution if conversion evaluated as H<sub>2</sub> to NO<sub>x</sub> ratio
- Synthetic mixture had different residence time, which affects both NO<sub>x</sub> conversion and un-selective H<sub>2</sub> reactions

### Significance

- Same catalyst evaluated at both lab scale and engine conditions
- Important comparison with lab scale results (on-going work)

### Conclusions

The EATS enables experimentation that is **not possible** using traditional methods.